

Watch the following video on Stoichiometry and then answer the following questions.

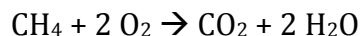
<https://www.youtube.com/watch?v=LQq203gyftA>

The video is also linked directly on the class website – loganchemistry.weebly.com.

1. What is stoichiometry? _____
2. Mr. Anderson relates stoichiometry to what type of game? _____
3. What do the coefficients tell us? _____
4. What should always be your first step in solving these problems?
5. What is a Limiting Reactant?
6. What is the percent yield equation (don't solve it, simply state it)?

Show your work for the following problems. Don't forget units!

7. If you have 24.6 g CH₄, how many grams of H₂O would you produce given the following reaction:



8. If you have 10.0 g CH₄ and 10.0 g O₂, how many grams of CO₂ is produced (still use reaction listed on the other side)?

9. What was the limiting reactant in this problem?